
This copy is for your personal, non-commercial use only.

If you wish to distribute this article to others, you can order high-quality copies for your colleagues, clients, or customers by [clicking here](#).

Permission to republish or repurpose articles or portions of articles can be obtained by following the guidelines [here](#).

The following resources related to this article are available online at www.sciencemag.org (this information is current as of September 1, 2011):

Updated information and services, including high-resolution figures, can be found in the online version of this article at:

<http://www.sciencemag.org/content/333/6047/1279.full.html>

Supporting Online Material can be found at:

<http://www.sciencemag.org/content/suppl/2011/08/31/333.6047.1279.DC1.html>

This article **cites 33 articles**, 9 of which can be accessed free:

<http://www.sciencemag.org/content/333/6047/1279.full.html#ref-list-1>

This article appears in the following **subject collections**:

Chemistry

<http://www.sciencemag.org/cgi/collection/chemistry>

6. J. Vermant, M. J. Solomon, *J. Phys. Condens. Matter* **17**, R187 (2005).
7. J. M. Brader, *J. Phys. Condens. Matter* **22**, 363101 (2010).
8. D. R. Foss, J. F. Brady, *J. Fluid Mech.* **407**, 167 (2000).
9. J. R. Melrose, R. C. Ball, *J. Rheol. (N.Y.N.Y.)* **48**, 961 (2004).
10. H. M. Laun *et al.*, *J. Rheol. (N.Y.N.Y.)* **36**, 743 (1992).
11. B. J. Maranzano, N. J. Wagner, *J. Chem. Phys.* **117**, 10291 (2002).
12. D. Kalman, N. J. Wagner, *Rheol. Acta* **48**, 897 (2009).
13. R. L. Hoffman, *J. Colloid Interface Sci.* **46**, 491 (1974).
14. R. L. Hoffman, *J. Rheol. (N.Y.N.Y.)* **42**, 111 (1998).
15. C. B. Holmes, M. E. Cates, M. Fuchs, P. Sollich, *J. Rheol. (N.Y.N.Y.)* **49**, 237 (2005).
16. D. Lootens, H. van Damme, Y. Hémar, P. Hébraud, *Phys. Rev. Lett.* **95**, 268302 (2005).
17. A. Fall, N. Huang, F. Bertrand, G. Ovarlez, D. Bonn, *Phys. Rev. Lett.* **100**, 018301 (2008).
18. E. Brown *et al.*, *Nat. Mater.* **9**, 220 (2010).
19. E. Brown, H. M. Jaeger, Preprint at <http://arxiv.org/abs/1010.4921> (2010).
20. Y. L. Wu, D. Derks, A. van Blaaderen, A. Imhof, *Proc. Natl. Acad. Sci. U.S.A.* **106**, 10564 (2009).
21. L. Isa, R. Besseling, A. B. Schofield, W. C. K. Poon, *Adv. Polym. Sci.* **236**, 163 (2010).
22. K. M. Schmoller, P. Fernández, R. C. Arevalo, D. L. Blair, A. R. Bausch, *Nat. Commun.* **1**, 134 (2010).
23. Materials and methods are available as supporting material on Science Online.
24. C. R. Nugent, K. V. Edmond, H. N. Patel, E. R. Weeks, *Phys. Rev. Lett.* **99**, 025702 (2007).
25. F. Parsi, F. Gadala-Maria, *J. Rheol. (N.Y.N.Y.)* **31**, 725 (1987).
26. C. Gao, S. D. Kulkarni, J. F. Morris, J. F. Gilchrist, *Phys. Rev. E Stat. Nonlin. Soft Matter Phys.* **81**, 041403 (2010).
27. J. F. Brady, G. Bossis, *J. Fluid Mech.* **155**, 105 (1985).
28. M. Urbakh, J. Klafter, D. Gourdon, J. Israelachvili, *Nature* **430**, 525 (2004).

Acknowledgments: We thank T. Beatus, Y.-C. Lin, J. Brady, L. Ristroph, and N. Wagner for useful discussions. This research was supported by grants from NSF Civil, Mechanical, and Manufacturing Innovation, Division of Materials Research (DMR), and DMR Materials Research Science and Engineering Centers, and in part by award KUS-C1-018-02 from King Abdullah University of Science and Technology (KAUST). J.N.I. was supported by the U.S. Department of Energy, Division of Materials Sciences and Engineering under award DE-FG02-87ER-45331.

Supporting Online Material

www.sciencemag.org/cgi/content/full/333/6047/1276/DC1

Materials and Methods

SOM Text

Figs. S1 to S7

Movies S1 to S6

References (29–40)

15 April 2011; accepted 15 July 2011

10.1126/science.1207032

Traffic Jams Reduce Hydrolytic Efficiency of Cellulase on Cellulose Surface

Kiyohiko Igarashi,^{1,*†} Takayuki Uchihashi,^{2,3,4,*} Anu Koivula,⁵ Masahisa Wada,^{1,6} Satoshi Kimura,^{1,6} Tetsuaki Okamoto,^{1,2} Merja Penttilä,⁵ Toshio Ando,^{2,3,4} Masahiro Samejima¹

A deeper mechanistic understanding of the saccharification of cellulosic biomass could enhance the efficiency of biofuels development. We report here the real-time visualization of crystalline cellulose degradation by individual cellulase enzymes through use of an advanced version of high-speed atomic force microscopy. *Trichoderma reesei* cellobiohydrolase I (*TrCel7A*) molecules were observed to slide unidirectionally along the crystalline cellulose surface but at one point exhibited collective halting analogous to a traffic jam. Changing the crystalline polymorphic form of cellulose by means of an ammonia treatment increased the apparent number of accessible lanes on the crystalline surface and consequently the number of moving cellulase molecules. Treatment of this bulky crystalline cellulose simultaneously or separately with *T. reesei* cellobiohydrolase II (*TrCel6A*) resulted in a remarkable increase in the proportion of mobile enzyme molecules on the surface. Cellulose was completely degraded by the synergistic action between the two enzymes.

Biorefining encompasses production of fuels, power, heat, and value-added chemicals by appropriate conversion of ligno-cellulosic feedstocks. This prospect offers many advantages, including diminished carbon dioxide emission, productive use of renewable nonfood crops or inedible waste products, and reduction of petroleum use. One of the bottlenecks to wide-

spread biorefining application is the enzymatic hydrolysis of the cellulosic raw material into sugars. In attempts to maximize efficiency and reduce costs, many studies on both individual cellulases and enzyme cocktails have been carried out (*1*). Various pretreatment methods have also been examined to increase the amount of sugar generated from cellulosic biomass with reduced enzyme loading and energy consumption, aiming at the development of commercially viable bioprocesses. It is generally recognized that one of the problems in cellulose hydrolysis is the slowdown of enzyme action with time and conversion (*2, 3*). Cellulose is a major component of plant cell walls, accounting for almost half of their net weight. Cell wall cellulose typically has ~70% crystallinity, suggesting that approximately one-third of net cellulosic biomass consists of natural crystalline cellulose, which is generally called cellulose I (*4*). Because cellulose chains have stable β -1,4-glucosidic bonds and each chain is also stabilized by intra- and intermolecular hydrogen bonds, cellulose I is quite resistant not only to

chemical hydrolysis but also to enzymatic degradation (*5*).

The industrially important cellulolytic ascomycete fungus *Trichoderma reesei* (anamorph of *Hypocrea jecorina*) secretes two extracellular cellobiohydrolases (CBHs), which are cellulases that can hydrolyze glycosidic linkages particularly at a crystalline surface, and form cellobiose (β -1,4-glucosidic dimer) as a major product from cellulose I (*6*). The two *T. reesei* CBHs hydrolyze crystalline cellulose from the reducing and non-reducing ends, respectively (*7*). These enzymes have a similar two-domain organization: The cellulose-binding domain (CBD), which is categorized into carbohydrate-binding module (CBM) family 1, contributes to adsorption on the insoluble substrate, and the cellulose-hydrolyzing catalytic domain (CD) catalyzes cleavage of glycosidic bonds. The two *T. reesei* cellobiohydrolase CDs have different types of folds. The CD of CBH I belongs to the glycoside hydrolase (GH) family 7, and the CD of CBH II belongs to the GH family 6, as listed on the Carbohydrate-Active enZyme (CAZy) server (*8*), and thus, the two enzymes are called *TrCel7A* and *TrCel6A*, respectively. In both cellobiohydrolases, the catalytic amino acids are located in a relatively long tunnel formed by surface loops extending from the central fold of the CD (*9, 10*). We previously visualized the linear movement of wild-type and isolated CD on cellulose I_α using high-speed atomic force microscopy (HS-AFM); the data suggested that the sliding movement of wild-type *TrCel7A* reflects the processive degradation of the cellulose chain by catalysis at the CD and requires initial recognition of the cellulose chain. The chain recognition involves the tryptophan residue W40 at the entrance of the *TrCel7A* active site tunnel (*11*).

In the present study, we succeeded in enhancing the temporal and spatial resolutions by using a laboratory-built HS-AFM with extensive improvements over the version reported previously (*12*), and we were able to visualize the movement of *TrCel7A* molecules on crystalline cellulose in detail. Enzyme molecules sliding

¹Department of Biomaterial Sciences, Graduate School of Agricultural and Life Sciences, University of Tokyo, Bunkyo-ku, Tokyo 113-8657, Japan. ²Department of Physics, Kanazawa University, Kanazawa 920-1192, Japan. ³Bio-AFM Frontier Research Center, College of Science and Engineering, Kanazawa University, Kakuma-machi, Kanazawa 920-1192, Japan. ⁴Core Research for Evolutional Science and Technology, Japan Science and Technology Agency, Sanbon-cho, Chiyoda-ku, Tokyo 102-0075, Japan. ⁵VTT Technical Research Centre of Finland, Post Office Box 1000, FI-02044 VTT, Finland. ⁶Department of Plant and Environmental New Resources, College of Life Sciences, Kyung Hee University, 1, Seocheon-dong, Giheung-ku, Yongin-si, Gyeonggi-do 446-701, Republic of Korea.

*These authors contributed equally to this work.

†To whom correspondence should be addressed. E-mail: aquarius@mail.ecc.u-tokyo.ac.jp (K.I.)

unidirectionally on the cellulose I_{α} surface are shown in Fig. 1A and movie S1. This movement of *Tr*Cel7A was observed only at the top of the cellulose crystal, where the individual enzyme molecules were moving as one line. The movement of the *Tr*Cel7A molecules was analyzed by using a custom software to track the linear movement of the center of each molecule, and the time course of molecular movement is shown in Fig. 1B. Some molecules slid continuously without stopping (demonstrated by the orange plot in Fig. 1B), whereas the movement of other *Tr*Cel7A molecules was often disturbed by halted and/or more slowly moving molecules (demonstrated by the open green and closed red plots in Fig. 1B), resembling the movement of traffic on a road. According to the histogram of the measured velocities (Fig. 1C), there would seem to be two populations of *Tr*Cel7A molecules having average velocities of -0.32 ± 3.4 and 7.1 ± 3.9 nm/s. This is consistent with the idea that *Tr*Cel7A molecules intermittently repeat stop and go movements.

It is generally believed that CBHs have two modes of adsorption on a cellulose surface: the productive adsorption mode, in which both the CD and CBD contribute to the binding, and the nonproductive adsorption mode, in which the enzyme molecules bind only via the CBD (13). In our previous HS-AFM study with *Tr*Cel7A and its mutants, that variant that bound only via the CBD (W40A without tryptophan residue at the active-site entrance) did not remain long on the surface and therefore could not be continuously visualized (11). In contrast, molecules bound to the substrate chain via the CD (the *Tr*Cel7A wild-type enzyme, isolated CD without CBD, and the inactive mutant E212Q) could be continuously visualized during HS-AFM observation. Considering these results, it seems likely that the *Tr*Cel7A molecules visualized here as stationary in the middle of cellulose fibrils have been stopped while holding a substrate chain in the active site tunnel of the CD. Moreover, we previously reported an average velocity of 3.5 ± 1.1 nm/s for the movement of *Tr*Cel7A (11), as shown by a dotted line in Fig. 1B. The velocity of moving molecules obtained in the present study was 7.1 ± 3.9 nm/s—twice the velocity estimated in the previous study—whereas the averaged velocity for all molecules (stopped and moving) was 5.3 ± 4.9 nm/s. Therefore, it seems likely that this earlier value (for average velocity) was determined by the two populations of moving and stopped molecules. Although the time-resolution of the HS-AFM observation used in this study was up to 300 ms/frame, greater than three times faster than that in the previous work (up to 1 s/frame), this time-resolution is still not high enough to analyze each hydrolytic step of *Tr*Cel7A. If one hydrolytic cycle produces one cellobiose unit, which is approximately 1 nm in length, each individual hydrolysis-related movement could occur within 140 ms, as calculated from the velocity obtained in the present study.

As already mentioned, cellulose I is quite resistant to enzymatic hydrolysis. We have shown recently that crystalline polymorphic conversion from cellulose I to cellulose III₁ through supercritical ammonia treatment dramatically enhances the hydrolysis of crystalline cellulose by *Tr*Cel7A (14). In order to investigate the reason for the enhanced hydrolysis of cellulose III₁, the degradation of the crystalline celluloses was visualized with HS-AFM, and the results were compared with those

of the conventional adsorption study. The binding isotherms—plots of free enzyme concentration versus amount of adsorbed enzyme ($[F]-A$)—followed a Langmuir-type two-binding site model on both crystalline celluloses (cellulose I_{α} and III₁) (Fig. 2A). These data show that *Tr*Cel7A has higher binding capacity to cellulose III₁ than to cellulose I_{α} . The simulated curves were divided into high-affinity (solid line) and low-affinity (dashed line) components, as shown in Fig. 2B. There was

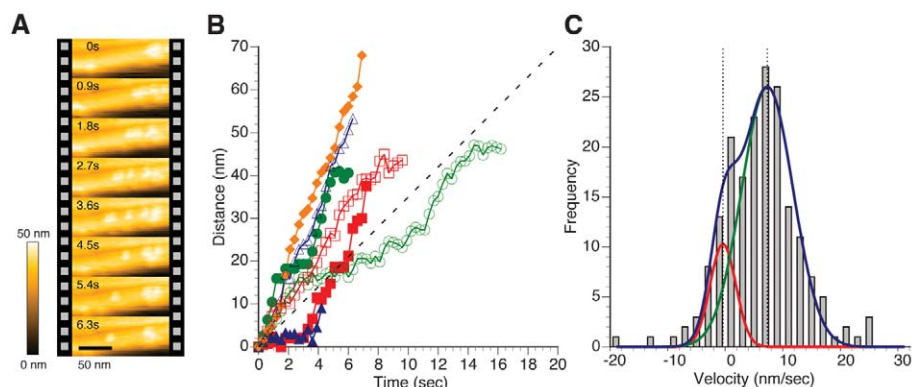


Fig. 1. (A) Real-time observation of crystalline cellulose I_{α} incubated with *Tr*Cel7A by means of HS-AFM. The time interval between images is 0.9 s. These images are taken from movie S1. (B) Time course of distance from the initial position for *Tr*Cel7A molecules. The mobility of each molecule was analyzed by using a routine developed in Igor Pro software, as described in the supporting online material. The different symbols indicate different molecules observed in HS-AFM images. The black dotted line represents the average velocity of *Tr*Cel7A (3.5 nm/s) estimated from previous HS-AFM observation (11). (C) Velocity distribution of the linear movement of *Tr*Cel7A on crystalline cellulose I_{α} ($n = 188$ xxxxx). The histogram was approximated by the combination of two Gaussian distribution curves (blue line) with mean \pm SD values of -0.32 ± 3.4 (red) and 7.1 ± 3.9 (green).

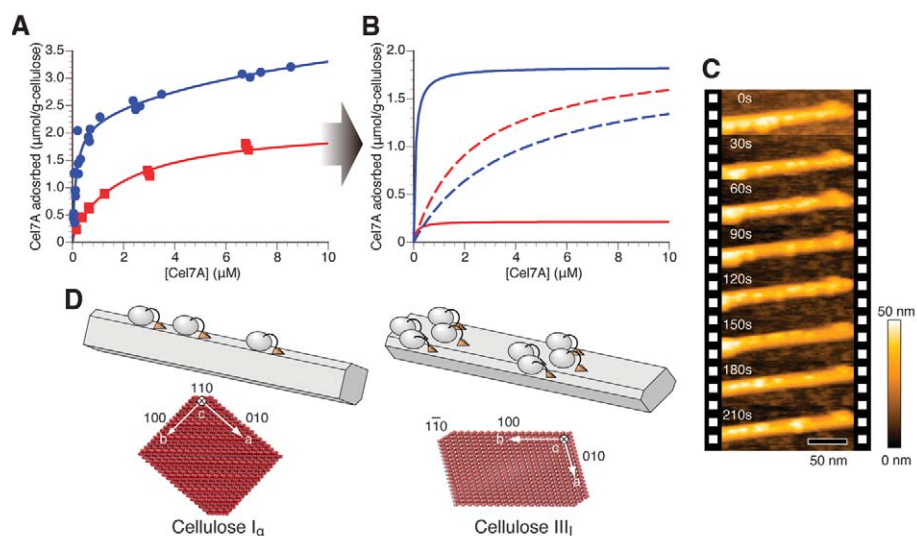
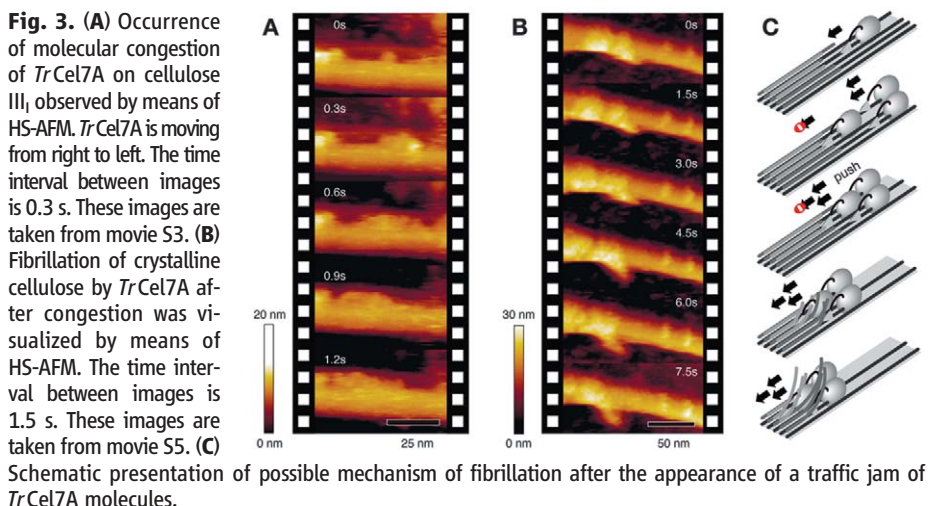


Fig. 2. (A) Free *Tr*Cel7A concentration dependence of the amount of adsorbed enzyme on crystalline cellulose I_{α} (red) and III₁ (blue). These plots were fitted to curves (red and blue solid lines) simulated by use of Langmuir's two-binding-site model, as described in (26); the high- (solid line) and low-affinity (dashed line) binding curves are individually drawn in (B). The binding parameters (A_1 , K_{ad1} , A_2 , and K_{ad2}) of *Tr*Cel7A for cellulose I_{α} were 0.22 ± 0.02 , 8.4 ± 1.4 , 2.0 ± 0.2 , and 0.44 ± 0.09 , and those for cellulose III₁ were 1.9 ± 0.1 , 11 ± 2 , 1.8 ± 0.3 , and 0.31 ± 0.06 (mean \pm SD), respectively. (C) HS-AFM images of crystalline cellulose III₁ degradation by *Tr*Cel7A. The time interval between images is 30 s. These images are taken from movie S2. (D) Schematic presentation of the increased number of observed lanes for *Tr*Cel7A on cellulose I_{α} (top left) compared with cellulose III₁ (top right) and comparison of sections of crystalline cellulose I_{α} (bottom left) and cellulose III₁ (bottom right) according to previous reports (17, 18).

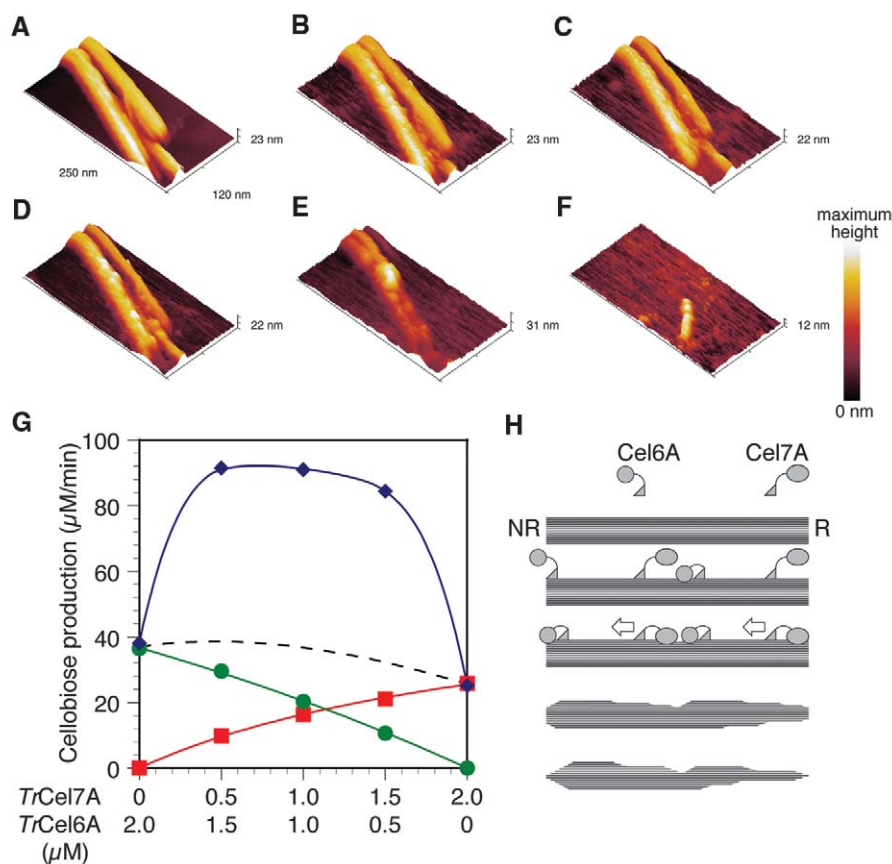
no apparent difference in the low-affinity component between the two crystalline polymorphs, whereas the high-affinity component in the case of crys-

talline cellulose III_I was quite large compared with that of crystalline cellulose I_α, which is consistent with our previous study (14).



When the hydrolysis of crystalline cellulose III_I by *Tr*Cel7A was observed with HS-AFM, as shown in Fig. 2C and movie S2, differences were apparent in the number but not in the velocity of enzyme molecules moving on cellulose III_I and cellulose I_α. In the case of crystalline cellulose I_α, enzyme molecules slid only along limited lanes (Fig. 1A and movie S1), whereas on cellulose III_I, *Tr*Cel7A molecules moved over almost the whole surface. This may reflect the difference in the amount of high-affinity binding (Fig. 2, A and B). Lehtiö and co-workers reported recently that the isolated CBD of *Tr*Cel7A binds to the 110 surface of *Valonia* cellulose (cellulose I_α), and they suggested that *Tr*Cel7A degrades crystalline cellulose from this hydrophobic surface (15). The hypothesis was recently confirmed by means of AFM observation (16). The characteristics of cellulose I_α and III_I differ (17, 18) in that cellulose I_α bears hydrophobic 110 surfaces between hydrophilic 100 and 010 surfaces, whereas cellulose III_I has similarly hydrophobic 110 and hydrophilic 010 surfaces but moderately hydrophobic (and larger) 100 surfaces, as shown in Fig. 2D. Because we used hydrophobic highly oriented pyrolytic graphite as a grid in the experiments, the crystals are expected to be oriented with their hydrophobic surfaces in contact with it and by symmetry on the opposite, top face as well. Therefore, it appears that *Tr*Cel7A moved not only on the hydrophobic 110 surface but also on the (somewhat less) hydrophobic 100 surface of cellulose III_I. In contrast, 110 is the only surface sufficiently hydrophobic for the enzyme to attach to in the case of cellulose I_α. Increasing the number of suitable lanes for cellulase movement would be an effective way to enhance crystalline cellulose hydrolysis, as judged from the comparison between cellulose I_α and cellulose III_I. In the case of *Tr*Cel7A action, moreover, when the movement of one molecule was halted many following molecules stacked behind it on the cellulose III_I surface, causing a traffic jam, as shown in Fig. 3A and movie S3. In Fig. 3B and movies S4 and S5, however, it becomes clear that after several additional molecules had also become blocked the enzyme molecules started to move again on the surface, and a cellulose bundle was peeled off from the crystalline cellulose. This behavior may reflect the presence of an obstruction on the surface of crystalline cellulose, which a single *Tr*Cel7A molecule is unable to climb over and is therefore halted. However, as schematically represented in Fig. 3C the accumulation of subsequent molecules behind the blocked molecule seems to lead to elimination of the obstacle (or obstacles). As a result, the blocked molecules restart linear movement from the point where the first molecule had stopped.

In the process of crystalline cellulose hydrolysis by *T. reesei*, two major cellobiohydrolases, *Tr*Cel7A and *Tr*Cel6A, synergistically contribute as described above (7, 19). We imaged the hydrolytic processes with crystalline cellulose III_I



(H) Proposed mechanism for the exo-exo synergy between *Tr*Cel6A and *Tr*Cel7A.

in the presence of both *Tr*-Cel6A and *Tr*-Cel7A. The crystals were first observed without enzyme (Fig. 4A and movie S6) then incubated *Tr*-Cel6A for 495 s (Fig. 4, B and C, and movies S7 and S8), and the same crystalline cellulose was again visualized after further addition of *Tr*-Cel7A (Fig. 4, D to F, and movies S7 and S9). As shown in Fig. 4, A to C, and movie S8, the appearance of the cellulose crystals did not change during incubation with *Tr*-Cel6A, although many enzyme molecules were observed on the surfaces. When *Tr*-Cel7A was added, enzyme molecules were seen to start moving from many points on the surface of the substrate, and the crystals apparently became thinner after the passage of the enzyme molecules, as shown in Fig. 4, D to F, and movie S9. This degradation of crystalline cellulose was dramatically faster than was the case with *Tr*-Cel7A alone (movie S2).

We also examined this synergetic action between the two enzymes biochemically. Cellobiose production from the reaction mixture containing both *Tr*-Cel6A and *Tr*-Cel7A (Fig. 4G, blue line) was greater than the sum (dotted line) of production by the individual enzymes separately (green and red lines for *Tr*-Cel6A and *Tr*-Cel7A, respectively), confirming the occurrence of synergistic hydrolysis of crystalline cellulose by these two enzymes, which is in agreement with previous findings. This has been called exo-exo synergy (20). These two enzymes both have active sites located in a tunnel that passes through the whole CD. The active site tunnel of *Tr*-Cel6A is formed by two surface loops and is about 20 Å long, whereas that of *Tr*-Cel7A is approximately 50 Å long, comprising six loops (9, 10, 21, 22). Previous transmission electron microscopic observations of crystalline cellulose after hydrolysis by *Tr*-Cel6A and *Tr*-Cel7A demonstrated that the two enzymes degrade the substrate from different ends of the crystal; *Tr*-Cel7A degrades the substrate from the reducing end, leading to fibrillation, thinning of the crystal, or narrowing of the crystal end,

whereas *Tr*-Cel6A hydrolyzes the cellulose chain from the nonreducing end less processively than *Tr*-Cel7A, sharpening the crystal tip (23, 24). It is somewhat difficult to find an explanation for this type of synergistic action on highly crystalline cellulose. One possibility is that the two loops forming the active site tunnel of *Tr*-Cel6A can open and allow generation of nicks in the middle of crystalline cellulose and that these nicks become starting and ending points for *Tr*-Cel7A activity. This mode of synergistic action (schematically shown in Fig. 4H) could also be called endo-exo synergy (25), although as explained above, *Tr*-Cel6A is generally defined as an exo-glucanase (cellobiohydrolase). The synergistic hydrolysis was also observed when *Tr*-Cel7A and *Tr*-Cel6A were incubated at the same time with cellulose III_I.

The present results suggest that the roughness of the crystalline cellulose surface leads to the formation of traffic jams of productively bound cellulases. Thus, flattening the surface, removing obstacles, and/or increasing the number of lanes, entrances, and exits by means of pretreatment or combined use of synergistically acting enzymes should reduce the molecular congestion, improving the mobility of the cellulase molecules and increasing the efficiency of hydrolysis.

References and Notes

- M. E. Himmel *et al.*, *Science* **315**, 804 (2007).
- J. Jalak, P. Väljamäe, *Biotechnol. Bioeng.* **106**, 871 (2010).
- B. Yang, D. M. Willies, C. E. Wyman, *Biotechnol. Bioeng.* **94**, 1122 (2006).
- D. N. S. Hon, *Cellulose* **1**, 1 (1994).
- R. Wolfenden, Y. Yuan, *J. Am. Chem. Soc.* **130**, 7548 (2008).
- T. T. Teeri, *Trends Biotechnol.* **15**, 160 (1997).
- T. T. Teeri *et al.*, *Biochem. Soc. Trans.* **26**, 173 (1998).
- B. Henrissat, A. Bairoch, *Biochem. J.* **293**, 781 (1993).
- C. Divne *et al.*, *Science* **265**, 524 (1994).
- J. Rouvinen, T. Bergfors, T. Teeri, J. K. Knowles, T. A. Jones, *Science* **249**, 380 (1990).
- K. Igarashi *et al.*, *J. Biol. Chem.* **284**, 36186 (2009).
- T. Ando *et al.*, *Proc. Natl. Acad. Sci. U.S.A.* **98**, 12468 (2001).

- J. Ståhlberg, G. Johansson, G. Pettersson, *Biotechnology (N. Y.)* **9**, 286 (1991).
- K. Igarashi, M. Wada, M. Samejima, *FEBS J.* **274**, 1785 (2007).
- J. Lehtö *et al.*, *Proc. Natl. Acad. Sci. U.S.A.* **100**, 484 (2003).
- Y. S. Liu *et al.*, *J. Biol. Chem.* **286**, 11195 (2011).
- M. Wada, H. Chanzy, Y. Nishiyama, P. Langan, *Macromolecules* **37**, 8548 (2004).
- Y. Nishiyama, J. Sugiyama, H. Chanzy, P. Langan, *J. Am. Chem. Soc.* **125**, 14300 (2003).
- L. E. R. Berghem, L. G. Pettersson, U. B. Axiofredriksson, *Eur. J. Biochem.* **53**, 55 (1975).
- B. Nidetzky, W. Steiner, M. Hayn, M. Claeysens, *Biochem. J.* **298**, 705 (1994).
- C. Divne, J. Ståhlberg, T. T. Teeri, T. A. Jones, *J. Mol. Biol.* **275**, 309 (1998).
- A. Koivula *et al.*, *J. Am. Chem. Soc.* **124**, 10015 (2002).
- H. Chanzy, B. Henrissat, *FEBS Lett.* **184**, 285 (1985).
- T. Imai, C. Boisset, M. Samejima, K. Igarashi, J. Sugiyama, *FEBS Lett.* **432**, 113 (1998).
- T. M. Wood, S. I. McCrae, *Biochem. J.* **171**, 61 (1972).
- K. Igarashi, M. Wada, R. Hori, M. Samejima, *FEBS J.* **273**, 2869 (2006).

Acknowledgments: The authors are grateful to K. Tokuyasu of the National Food Research Institute, J. Ståhlberg of the Swedish University of Agricultural Sciences, and A. Isogai of the University of Tokyo for their critical suggestions during the preparation of this paper. We thank T. Tsukada for his help in checking the activity and purity of *Tr*-Cel6A. This research was supported by Grants-in-Aid for Scientific Research to K.I. (19688016 and 21688023), T.U. (21023010 and 21681017), and T.A. (20221006) from the Japanese Ministry of Education, Culture, Sports, and Technology; by a grant of the Knowledge Cluster Initiative to T.A.; by a Grant for Development of Technology for High Efficiency Bioenergy Conversion Project to M.S. (07003004-0) from the New Energy and Industrial Technology Development Organization; and by a Grant for Development of Biomass Utilization Technologies for Revitalizing Rural Areas to M.S. from the Japanese Ministry of Agriculture, Forestry and Fisheries.

Supporting Online Material

www.sciencemag.org/cgi/content/full/333/6047/1279/DC1
Materials and Methods
References (27–33)
Movies S1 to S9

16 May 2011; accepted 29 July 2011
10.1126/science.1208386

Isotopic Signature of N₂O Produced by Marine Ammonia-Oxidizing Archaea

Alyson E. Santoro,^{1*} Carolyn Buchwald,² Matthew R. McIlvin,¹ Karen L. Casciotti^{1*†}

The ocean is an important global source of nitrous oxide (N₂O), a greenhouse gas that contributes to stratospheric ozone destruction. Bacterial nitrification and denitrification are thought to be the primary sources of marine N₂O, but the isotopic signatures of N₂O produced by these processes are not consistent with the marine contribution to the global N₂O budget. Based on enrichment cultures, we report that archaeal ammonia oxidation also produces N₂O. Natural-abundance stable isotope measurements indicate that the produced N₂O had bulk δ¹⁵N and δ¹⁸O values higher than observed for ammonia-oxidizing bacteria but similar to the δ¹⁵N and δ¹⁸O values attributed to the oceanic N₂O source to the atmosphere. Our results suggest that ammonia-oxidizing archaea may be largely responsible for the oceanic N₂O source.

Tropospheric concentrations of nitrous oxide (N₂O) are currently 322 parts per billion and rising at a rate of ~0.25% per year

(1). Future projections for this radiatively active trace gas are uncertain, primarily because biological sources and sinks of N₂O are spatially and

temporally variable, and their mechanisms are not fully understood. Marine N₂O sources to the atmosphere are estimated to represent ~30% of total “natural” inputs, or ~4 Tg N₂O-N per year (2). Changes in the magnitudes of these sources depend on the mechanisms and controls of microbial N₂O production.

Marine N₂O production is thought to be carried out by ammonia-oxidizing bacteria (AOB)

¹Department of Marine Chemistry and Geochemistry, Woods Hole Oceanographic Institution, Woods Hole, MA 02543, USA.

²Joint Program in Chemical Oceanography, Massachusetts Institute of Technology–Woods Hole Oceanographic Institution, Woods Hole, MA 02543, USA.

*To whom correspondence should be addressed. E-mail: asantoro@umces.edu (A.E.S.); kcasciotti@stanford.edu (K.L.C.)

†Present address: Horn Point Laboratory, University of Maryland Center for Environmental Science, Cambridge, MD 21613, USA.

‡Present address: Department of Environmental Earth System Science, Stanford University, Stanford, CA 94305, USA.